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Cardiac glycosides and other factors influencing  $\text{Na,K-ATPase}$  activity and the ratio between intracellular  $\text{Na}^+$  and  $\text{K}^+$  concentrations lead to changes in cardiac and smooth muscle tone [5, 9], secretion of hormones and mediators [6, 10], and sensitivity of tissues to the action of various hormones [11]. At the same time, interaction between actin and myosin, the level of secretory activity, and activity of enzymes of the cyclic nucleotide system are known to be regulated by the free intracellular  $\text{Ca}^{++}$  concentration [3, 13, 14]. Accordingly it can be tentatively suggested that in the cases mentioned above the influence of cardiac glycosides on tissue metabolism is mediated through a change in the intracellular distribution of  $\text{Ca}^{++}$  and, in particular, through a change in  $\text{Ca}^{++}$  accumulation by intracellular structures.

In the investigation described below the effect of the ratio between  $\text{Na}^+$  and  $\text{K}^+$  concentrations on  $\text{Ca}^{++}$  accumulation by isolated mitochondria of myocardium, brain, and adipose tissue, as the principal Ca-buffer factor in the cells of these tissues, was investigated [1, 12].

#### EXPERIMENTAL METHOD

Male Wistar rats weighing 150–180 g, aged 2–3 months, were used. The procedures for isolating mitochondria from myocardium, brain, and adipose tissue and also the characteristics of the membrane preparations obtained were described previously [1, 12, 15]. The method of determining incorporation of  $^{45}\text{Ca}$  into isolated membrane fractions also was described previously [1]. The composition of the incubation medium is given in the captions to Figs. 1–4.

Reagents:  $\text{NaCl}$ ,  $\text{KCl}$ ,  $\text{CaCl}_2$ , succinic acid, Tris-HCl (from Merck, West Germany), ATP-Tris (from Sigma, USA), ruthenium red (from BDH, England),  $^{45}\text{CaCl}_2$  (from the Radiochemical Centre, Amersham, England). The remaining reagents were from Soyuzreakhim, of the chemically pure grade.

#### EXPERIMENTAL RESULTS

The kinetics of  $\text{Ca}^{++}$  accumulation by myocardial mitochondria in the presence of ATP and succinate is shown in Fig. 1. Under these conditions, an equilibrium distribution of  $^{45}\text{Ca}$  was established between the mitochondria and incubation medium after 20 min of incubation. The Ca-accumulating capacity of the mitochondria, i.e., the maximal amount of  $\text{Ca}^{++}$  accumulated by mitochondria in incubation medium of this composition, was evidently determined by the ratio between the velocities of inflow and outflow of the cation.

It will be clear from Fig. 2 that an increase in the  $\text{NaCl}$  concentration to 60 mM accompanied by a decrease in the  $\text{KCl}$  concentration from 120 to 60 mM leads to a significant decrease in the Ca-accumulating capacity of the myocardial and brain mitochondria. Differences in the  $\text{Na}^+$  concentration required to obtain half the maximal value of the effect (7 and 30 mM for myocardial and brain mitochondria, respectively), incidentally, agreed with

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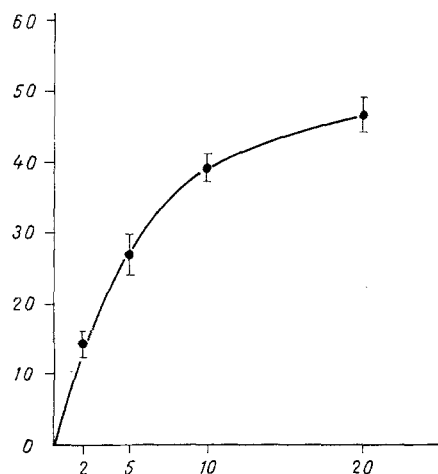


Fig. 1. Kinetics of  $\text{Ca}^{++}$  accumulation by myocardial mitochondria in incubation medium containing 50 mM Tris-HCl, pH 7.4; 120 mM KCl; 4 mM  $\text{MgCl}_2$ ; 12  $\mu\text{M}$   $\text{CaCl}_2$ ; 2  $\mu\text{Ci/ml}$  of  $^{45}\text{CaCl}_2$ ; 4 mM ATP-Tris, and 10 mM succinate-Tris. Abscissa, incubation time (in min); ordinate, Ca-accumulating capacity (in moles/mg protein).

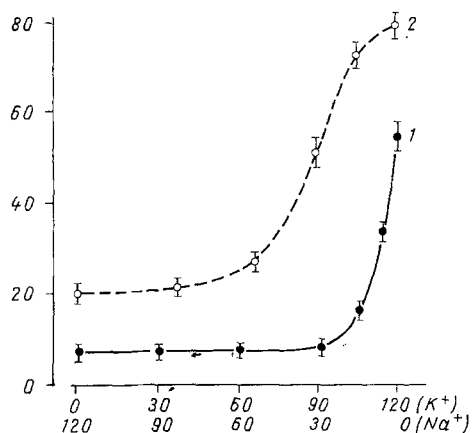


Fig. 2. Dependence of Ca-accumulating capacity of myocardial (1) and brain (2) mitochondria on ratio between  $\text{Na}^+$  and  $\text{K}^+$  concentrations in incubation medium containing 50 mM Tris-HCl, pH 7.4; 5 mM  $\text{MgCl}_2$ ; 12  $\mu\text{M}$   $\text{CaCl}_2$ ; 4 mM ATP-Tris, and 10 mM succinate-Tris. Abscissa, ratio between  $\text{K}^+$  and  $\text{Na}^+$  concentrations (in mM); ordinate, Ca-accumulating capacity (in nmoles/mg protein/20 min).

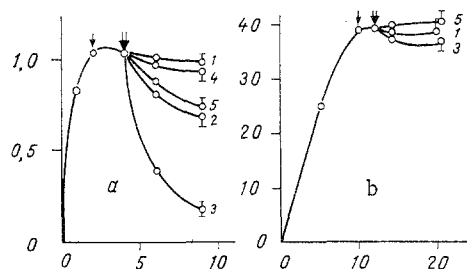


Fig. 3. Effect of monovalent cations on  $\text{Ca}^{++}$  liberation from mitochondria. Incubation medium: a) 50 mM Tris-HCl, pH 7.4; 250 mM sucrose, 5 mM  $\text{MgCl}_2$ , 12  $\mu\text{M}$   $\text{CaCl}_2$ , 2  $\mu\text{Ci/ml}$   $^{45}\text{CaCl}_2$ , and 10 mM Tris-succinate; b) the same + 4 mM ATP-Tris. Single arrow — ruthenium red (2.5  $\mu\text{M}$ ) added, double arrow — NaCl (2 and 3) or KCl (4 and 5) added in concentration of 10 mM (2 and 4) or 40 mM (3 and 5). 1) Control (without addition of monovalent cations). Remainder of legend as in Fig. 1.

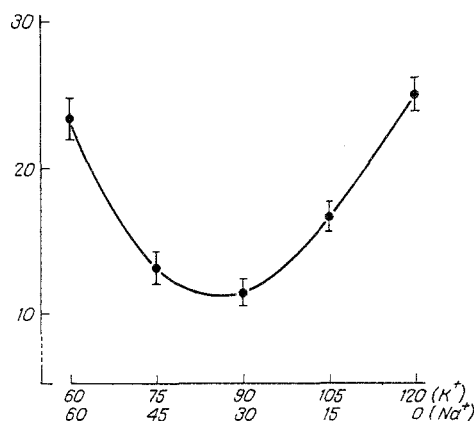


Fig. 4. Ca-accumulating capacity of mitochondria of adipose tissue as a function of ratio between  $\text{Na}^+$  and  $\text{K}^+$  concentrations in incubation medium. Composition of incubation medium and explanation of axes of coordinates given in caption to Fig. 2.

differences in the  $\text{Na}^+$  concentrations in the cytoplasm of these tissues (8-15 and 30-40 mM respectively for myocardium and nerve tissue [4]). These observations indicate that the Na-dependent regulation of the Ca-capacity of mitochondria of excitable tissues *in vivo* is a possibility.

The cause of the decrease in the Ca-accumulating capacity of the mitochondria with an increase in the  $\text{Na}^+$  concentration in the incubation medium could be slower inflow of  $\text{Ca}^{++}$  into and (or) faster outflow from the mitochondria. Data on the effect of monovalent cations on the electrochemical potential of the inner mitochondrial membrane, which determines the rate of  $\text{Ca}^{++}$  accumulation, are negative in character [2]. Meanwhile it was shown in 1974 that addition of  $\text{Na}^+$  leads to the rapid outflow of  $^{45}\text{Ca}$  from myocardial mitochondria previously loaded with isotope in the presence of succinate, and the further entry of  $\text{Ca}^{++}$  into which was retarded by the addition of ruthenium red [7, 8]. This effect was reproduced in the present experiments on the fraction of myocardial mitochondria (Fig. 3a). However, if the

mitochondria were loaded with  $^{45}\text{Ca}$  in the presence not only of succinate, but also of ATP, subsequent addition of  $\text{Na}^+$  did not lead to any significant release of  $\text{Ca}^{++}$  from the organelles (Fig. 3b).

The data given in Figs. 2 and 3a, b can be explained on the assumption that  $\text{Ca}^{++}$ -transport centers inside the mitochondria and the sites of the Na-Ca-carrier responsible for outflow of  $\text{Ca}^{++}$  are located close together on the inner membrane of the organelles, whereas the centers for formation of Ca-ATP-inorganic phosphate formation are distant from these sites and that the  $\text{Ca}^{++}$  present in these complexes plays no part in transmembrane exchange. The model put forward assumes that if ATP is present in the incubation medium, a Na-effect can be recorded only if the  $\text{Ca}^{++}$  inflow channels are uninhibited, which is in agreement with the experimental data described above.

It must be pointed out, however, that the mechanism of involvement of  $\text{Na}^+$  in regulation of the Ca-capacity of the mitochondria we have just examined may perhaps not be universal for mitochondria of all tissues. For instance, in the case of mitochondria of adipocytes, dependence of Ca-accumulating capacity on ratio between  $\text{Na}^+$  and  $\text{K}^+$  concentrations in the incubation medium is highly complex in character (Fig. 4): An increase in the  $\text{Na}^+$  concentration up to 30-40 mM reduced  $\text{Ca}^{++}$  accumulation by half, but further replacement of  $\text{K}^+$  by  $\text{Na}^+$  restored the Ca capacity of the organelles to the original values. Explanation of the mechanism of regulation of the Ca-accumulating capacity of the mitochondria of unexcitable tissues by monovalent cations evidently requires additional experimental study.

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